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Unit 4 Nuclear Chemistry Review Study Guide. 1. Define the following terms below. a) Half-life- the length of time during which half of a given number of atoms of a radioactive nuclide decay b) Nuclear Fission- the process in which lightweight nuclei combine to form heavier, more stable nuclei c) Nuclear Fusion- process in which a very heavy nucleus splits to form medium-weight nuclei 2.

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Chapter 25 - Nuclear Chemistry. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Ieslielaland. Study Guide for Chapter 25. Terms in this set (37) Neutron Ejection. when a neutron is emitted from the nucleus. ${}^1_0\text{n}$. Particle for Neutron Ejection. ${}^5_2\text{He} \rightarrow {}^1_0\text{n} + {}^4_2\text{He}$. Example of Neutron Ejection.

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Chapter 24 Study Guide Nuclear Chemistry Answers

Chapter 18 - Nuclear Chemistry 289 Key Ideas Answers 14. Because protons and neutrons reside in the nucleus of atoms, they are called nucleons. 16. There are two forces among the particles within the nucleus. The first, called the electrostatic force, is the force between electrically charged particles. The second force,

Chapter 18 Nuclear Chemistry

Circle the letter of the choice that best completes the statement. 1. The number of stable isotopes

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that exist compared to the number of unstable isotopes is a much less. b. much more. c. slightly more. d. about the same. 2. A lightweight isotope is likely to be stable if the ratio of protons to neutrons in its nucleus is a. 1:2. b 1:1. c. 2:1. c. beta particle.

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